

## Recovery and Application of Waste Carbon Fibre Reinforced Plastics (CFRP)

Jude Onwudili

Email: [j.onwudili@aston.ac.uk](mailto:j.onwudili@aston.ac.uk)



---

**Biomass and Waste/Fuel Characterisation, upgrading and Carbonisation Interest  
Group Seminar and Annual General Meeting**

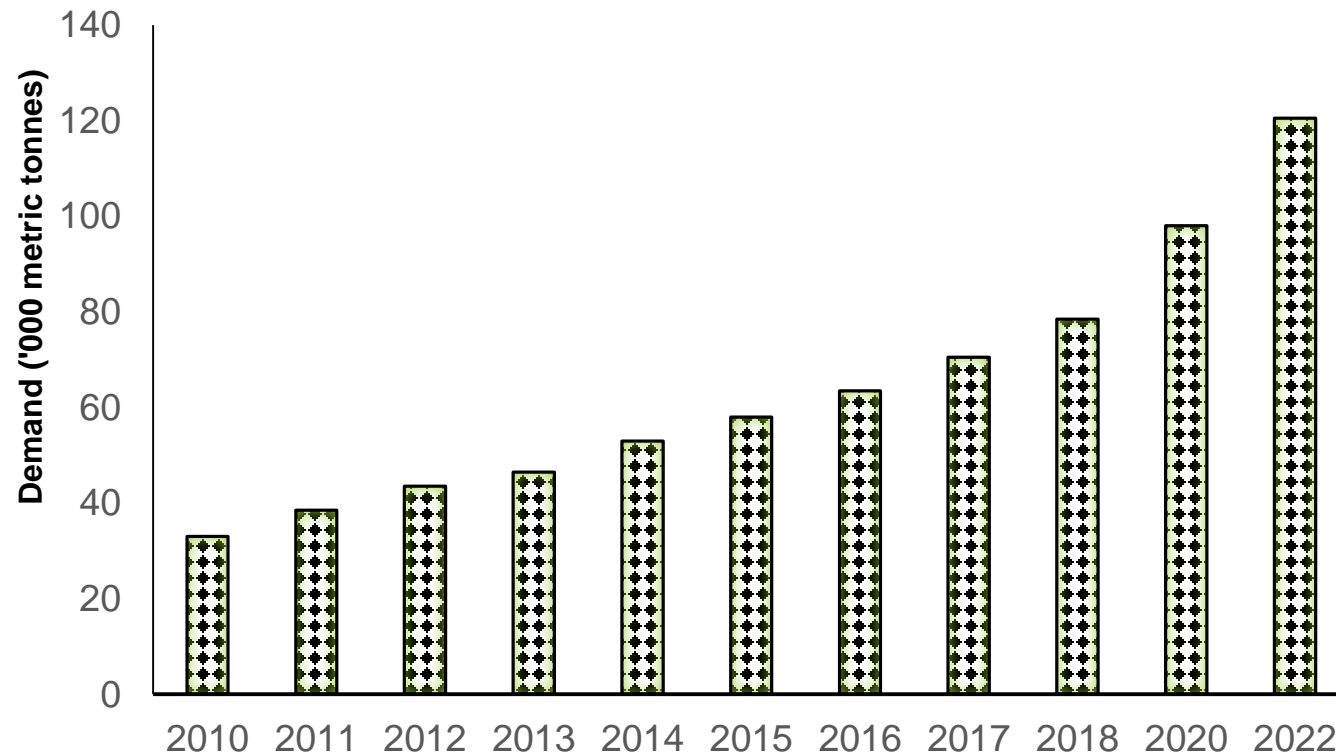
**Wednesday 10<sup>th</sup> April 2019**

**Aston University, Birmingham, UK**

- Introduction
- Aim & Objectives
- Materials and Methods
- Results from carbon fibre recovery using solvothermal media
- Results from composites made from of recovered carbon fibre
- Conclusions

# Introduction

The worldwide demand for carbon fibre was 38000 tonnes in 2010 and expected to rise to 125000 tonnes by 2020<sup>1</sup>



1. Tony, R., 2011. The Carbon Fibre Industry Worldwide 2011-2020: An Evaluation Of Current Markets And Future Supply And Demand. Materials Technologies Publications.

# Properties of Carbon Fibre

Carbon fibre is made of honeycomb-shaped carbon crystals aligned in a long axis in flattened ribbons (mainly from PAN (90%) and rayon)

- ✓ Fire Resistance/Not flammable
- ✓ High Thermal Conductivity in some forms
- ✓ Low coefficient of thermal expansion
- ✓ Non-toxic
- ✓ Biologically inert
- ✓ X-Ray Permeable
- ✓ Relatively Expensive
- ✓ High Strength to weight ratio
- ✓ Rigidity
- ✓ Corrosion resistance
- ✓ Electrical Conductivity
- ✓ Fatigue Resistance
- ✓ Good tensile strength but Brittle

# Uses and CFRP Waste Generation

- On a Boeing 787, 50% of the aircraft is composite, with 20% aluminium, 15% titanium, 10% steel and 5% other materials.
- About 60-70% of the structural weight of Formula One cars (e.g. McLaren Formula 1 – A)
- Other aerospace e.g. satellite bodies
- Sports equipment – tennis rackets, golf clubs
- Between 6000 to 8000 airplanes expected to reach their end of service life, which will generate CFRP wastes, by 2030<sup>2</sup>



Currently, about 3000 tonnes of CRFP waste is generated annually in Europe and UK.

Landfilling is not a good option – loss of valuable material and carbon fibre is non-putrescible

Incineration is not a good option – carbon fibre is flame retardant, hence large quantities of fuel would be required for its incineration

Often used as reinforcements for plastics with thermosetting resins as binding agents

Reuse and recycling appear to be viable solutions

# Main Routes for Recycling

Mechanical – bulk ball milling + air-classification

Thermal – mostly combustion and pyrolysis; removes resins and plastics for energy recovery

Chemical – to remove resins and plastics

- Solvolysis (hydrolysis, Glycolysis, Alcoholysis)

## Aims

- Recovery of carbon fibres from the waste with minimal impact on physical properties for re-application
- Recovery of the resin as monomers (valuable chemical feedstock) or as fuel

## Objectives

- The effect of solvents, temperature, type of catalyst and reaction time on resin removal efficiency
- Manufacture of composite materials and testing them



CFRP wastes supplied by ELG Carbon Fibre Ltd

## Elemental Composition of Waste CFRP

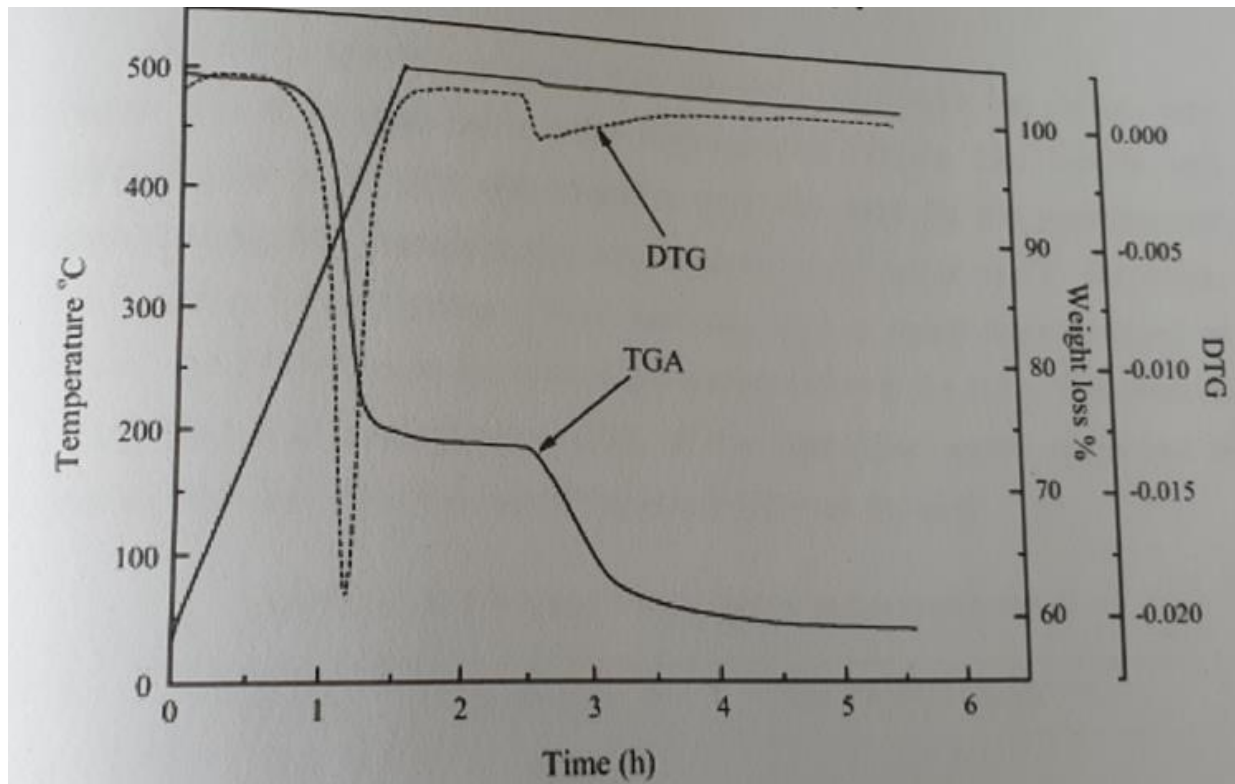
CFRP waste	C	N	H	S	O
	80.3	4.15	2.05	1.65	6.9

Water and Ethylene Glycol (EG) used as solvent

Solvothermal experiments at temperatures 380-400 °C

**Annually, about 2.6 billion litres of waste ethylene glycol is generated in the US alone; a large proportion of this comes from de-icing of **aeroplanes** and from spent heat transfer fluids**

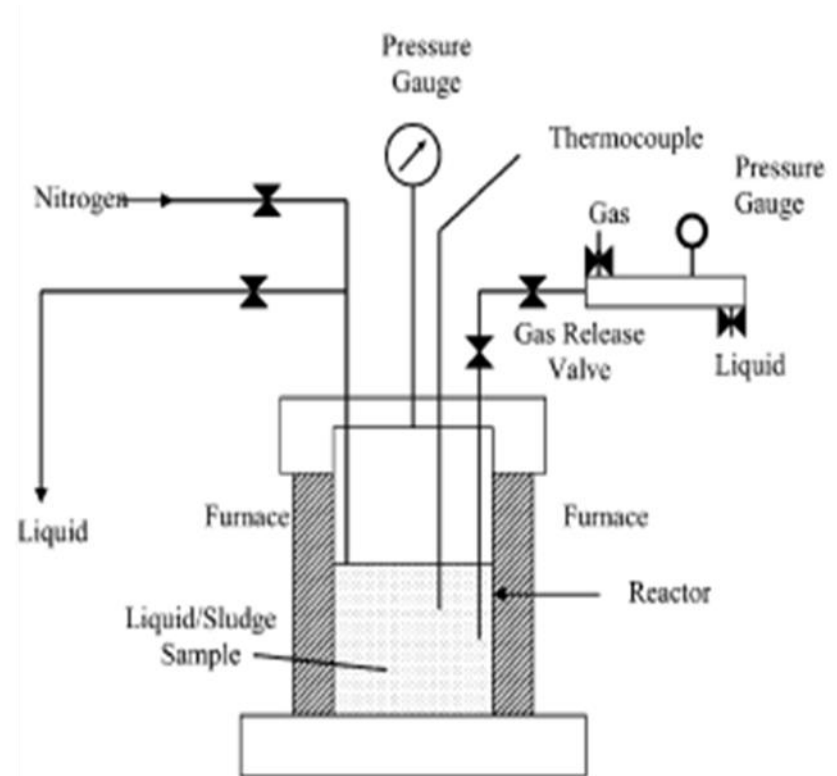
# Characterisation of Waste CFRP



- ❖ On average, CFRP consisted of 38.5 wt% of resin and 61.5 wt% of carbon fibre by TGA characterisation and ASTM D3174 – 12 ashing method.
- ❖ Virgin carbon fibre contained 99.5 wt% carbon fibre

# Experimental

- SS316 batch reactor obtained from the Parr Instrument Co., USA,
- 10 g of CFRP in 60 ml of solvent (EG/water)
- Gas analysis in Varian CP-3380 (GC/TCD & GC/FID)
- Liquid analysis in TOC for total carbon content.
- Solid characterisation

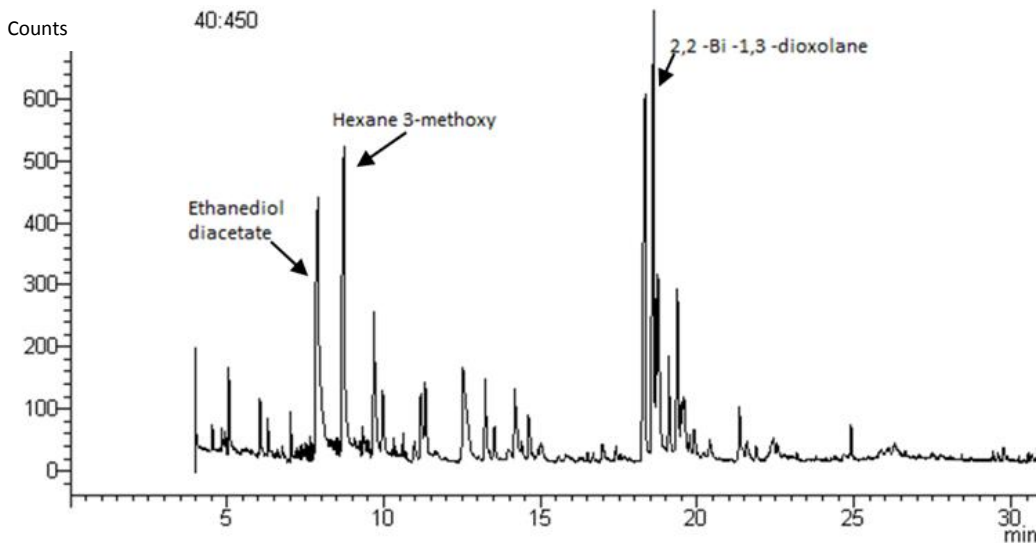


# Product distribution with H<sub>2</sub>O/EG systems

Sample No	H <sub>2</sub> O, mL	EG, mL	T, °C	*Liquid, %	Gas, %	Solid, %	Resin removal %
1	0	60	380	88.7	2.16	9.05	79.30
2	0	60	400	88.4	3.16	8.41	92.1
3	10	50	380	87.4	4.22	8.42	94.2
4	10	50	400	89.2	2.58	8.25	97.6
5	15	45	400	86.3	5.31	8.44	95.2
6	30	30	400	84.4	5.88	9.75	73.8
7	45	15	400	84.0	5.68	10.4	66.5
8	50	10	400	88.1	1.30	10.6	67.3
9	60	0	400	89.1	0.69	11.6	49.0

\*weight % of solvent(s) + liquid products by difference

# Organic products in liquid residuals



Organic compounds detected with ethylene glycol as solvent		
S/N	Compound	Structure
1	1,1-Ethanediol diacetate	<chem>CC(=O)OCCOCC(=O)C</chem>
2	2-Methyl-1,3-Dioxolane	<chem>CC1COCCO1</chem>
3	Ethanol, 2,2'-[1,2-ethanediylbis(oxy)] bis, diacetate	<chem>CC(=O)OCCOCCOCCOCCOCC(=O)C</chem>
4	2-(1-Methylethyl)-1,3-Dioxolane,	<chem>CC(C)C1COCCO1</chem>
5	Ethanol, 2,2'-oxybis-, diacetate	<chem>CC(=O)OCCOCCOCCOCC(=O)C</chem>
6	Ethanol, 2,2'-oxybis-, dipropionate	<chem>CCC(=O)OCCOCCOCCOCC(=O)CC</chem>
7	Hydroperoxide, 1,4-dioxan-2-yl	<chem>CC1COCCOCC1OO</chem>
8	2,2'-Bi-1,3-dioxolane	<chem>C1COCCO1C2COCCO2</chem>

# Mechanical properties of virgin and recovered fibre

	Recovered CF		
	*Virgin	1	2
Breaking Force, N	0.135	0.138	0.131
Elongation, mm	0.3	0.28	0.43
Tensile Strength, GPa	3.5	3.6	3.4
Young Modulus (MPa)	233	254	159

1 Recovered at 400°C, with 10 ml of H<sub>2</sub>O, 50 ml of EG

2 Recovered at 400°C, with EG only

\*Sourced in UK

Commercial LDPE (Bralen RB 2-62, Tisza Chemical Group Public Limited Company, Hungary)

## Virgin Fibre

Properties	PANEX®33
Tensile strength, MPa	3800
Tensile modulus, GPa	228
Density, g/cm <sup>3</sup>	1,81
Diameter, mm	7,2
Carbon content, %	95
Sizing	no

## LDPE

Properties	Bralen RB 2-62
Tensile Strength, Mpa	11.4
Tensile Modulus, MPa	348
Tensile Extension, %	155
Flexural Strength, MPa	7.5
Flexural Modulus,MPa	495

Commercial additives (coupling agents):

- 3 – Amino – Propyl - Silane (silane-based additive)
- Grafted – Maleic Anhydride

# Composite Making

Two roll mill (Labtech Ltd, Thailand)

- First mill at 180 °C; second mill at 150 °C
- Press moulding at 180 °C



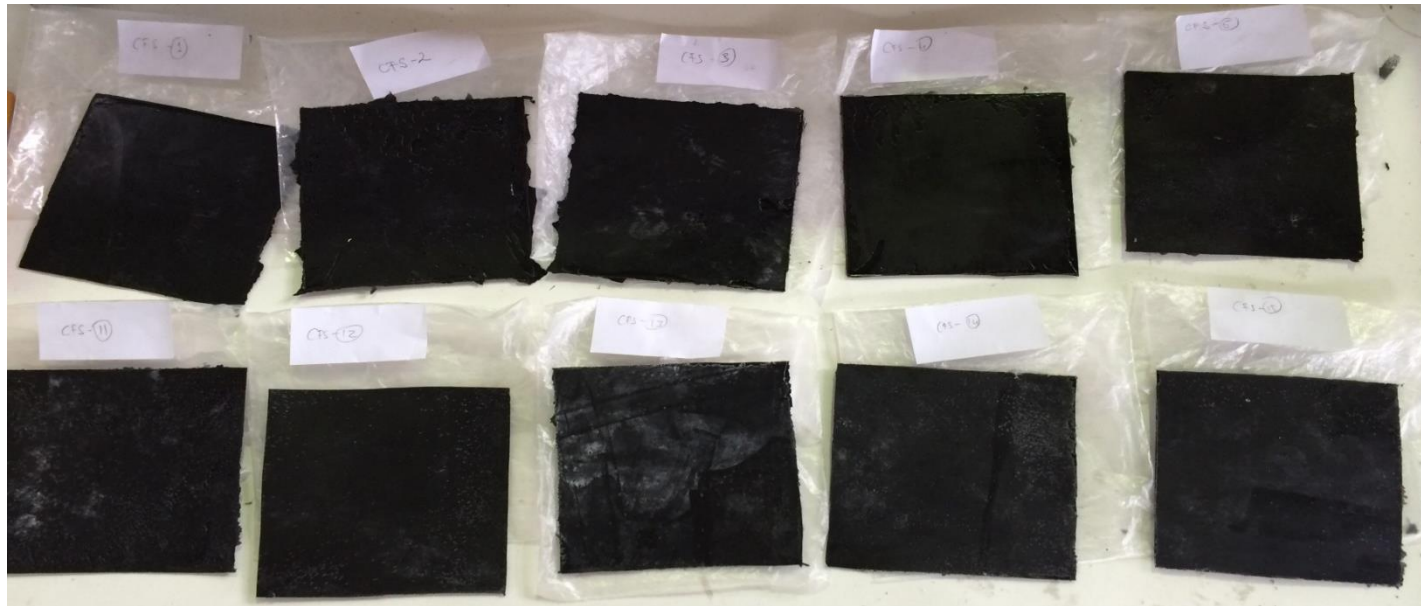


Ten samples of composite materials were prepared as follows:

	*Composite with Recovered Fibre					Composite with Virgin Fibre				
Compound (wt %)	CFS-1	CFS-2	CFS-3	CFS-4	CFS-5	CFS-11	CFS-12	CFS-13	CFS-14	CFS-15
LDPE	85	84	84	83	83	85	84	84	83	83
Recovered CF	15	15	15	15	15	-	-	-	-	-
Virgin CF	-	-	-	-	-	15	15	15	15	15
Silane based	-	1	-	-	-	-	1	-	-	-
grafted-MA	-	-	1	-	-	-	-	1	-	-
Ad-1	-	-	-	2	-	-	-	-	2	-
Ad-4	-	-	-	-	2	-	-	-	-	2

\* CF recovered at 400 °C, with 10 mL of H<sub>2</sub>O, 50 mL of EG

# Composites Produced



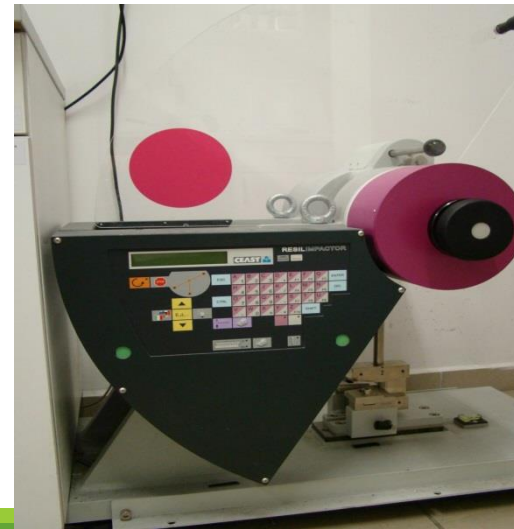
# Testing of Composites

Tensile properties on Instron 3345 testing machine (according to MSZ EN ISO 527-1-4:199 std)

Composites were cut into 10 x 1 x 100 mm

Charpy test on CEAST Resil Impactor according to MSZ EN ISO 179-2:2000 std mtd

Samples 6.5 mm deep V-form notch cut



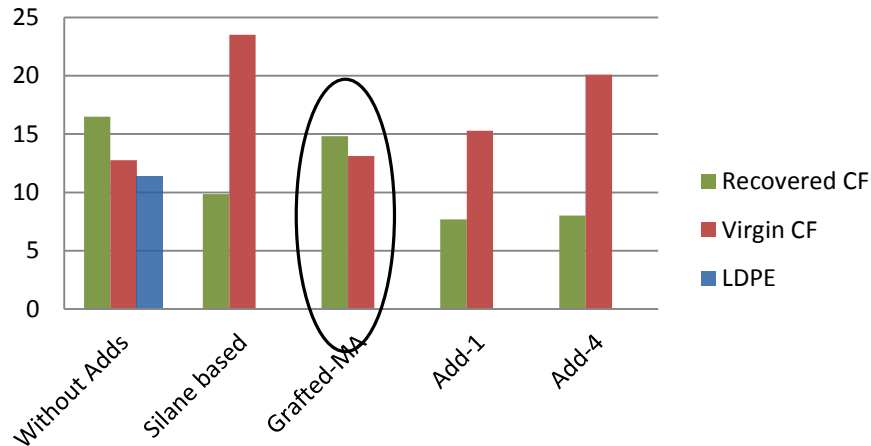
# Mechanical properties of virgin and recovered fibre

- Breaking Force (N)
- Tensile Strength (GPa)
- Tensile Strain extension at maximum load (mm)
- Young's modulus (MPa) – stiffness test
- Flexural Properties
- Charpy (Impact Energy) Test

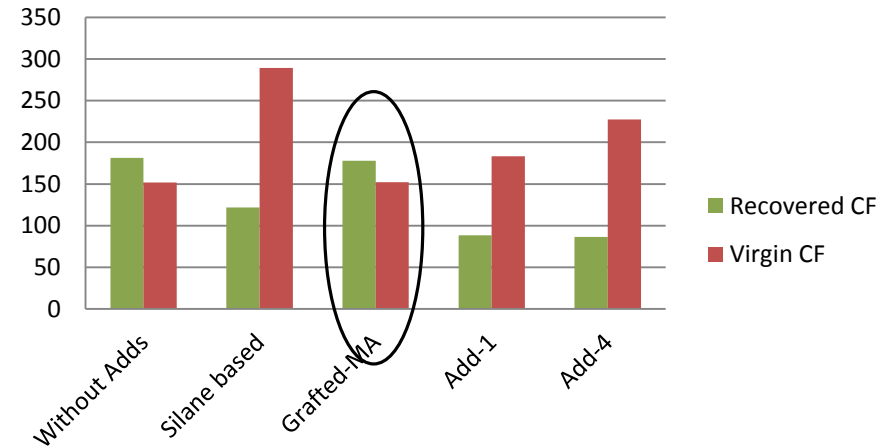
Also tested effect of the additives to the mechanical properties

# Tensile Tests Results from Composites

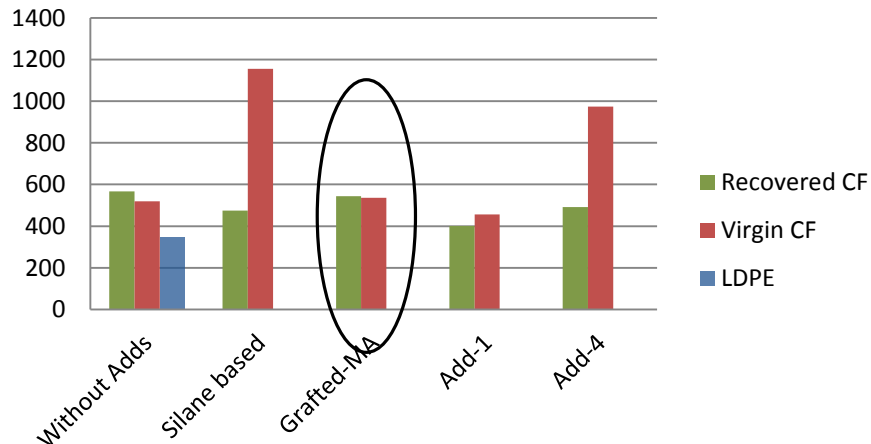
## Tensile Strength (MPa)



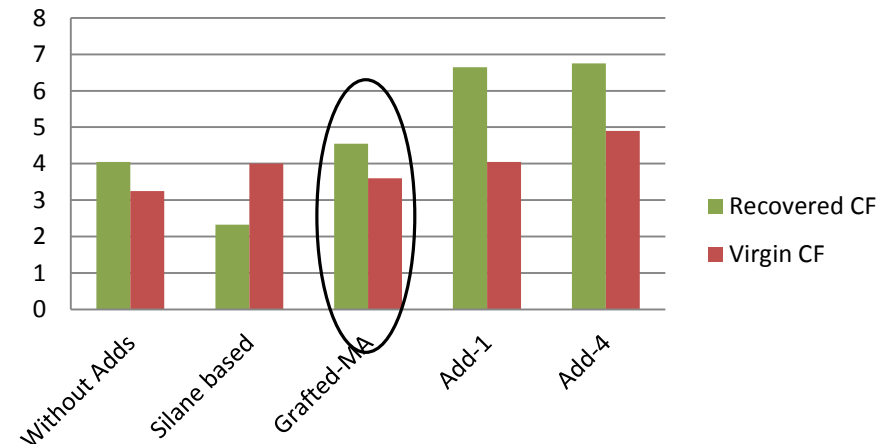
## Load at Max. Tensile Extension (N)



## Young's Modulus (MPa)

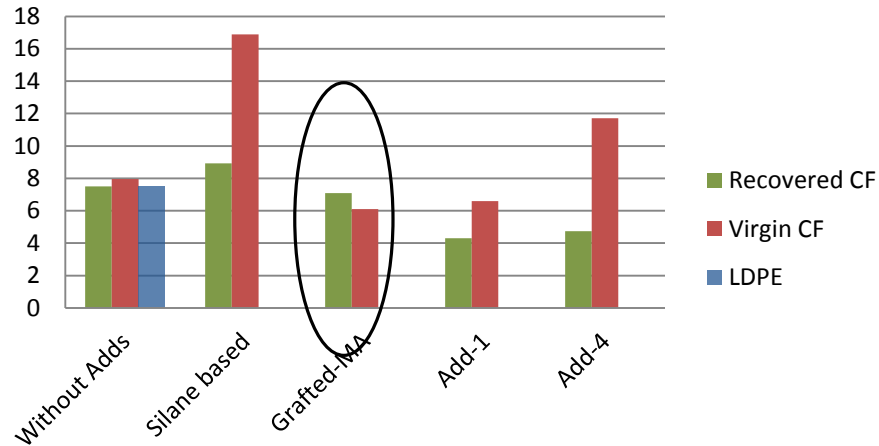


## Extension at Max. Load (mm)

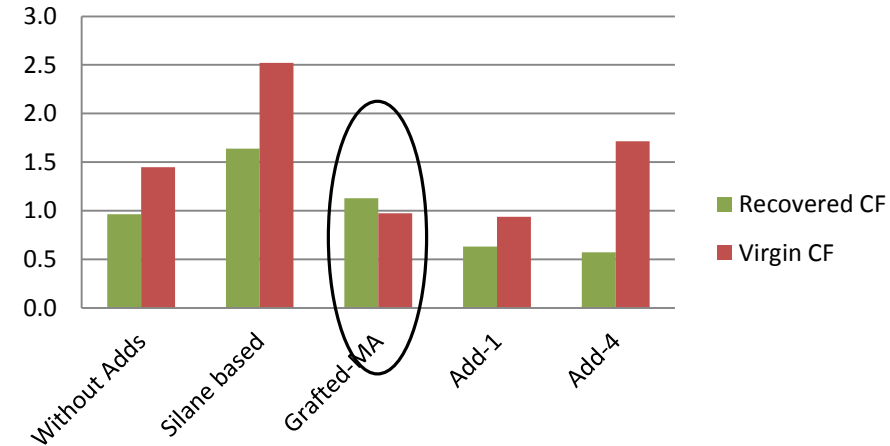


# Flexural Tests Results from Composites

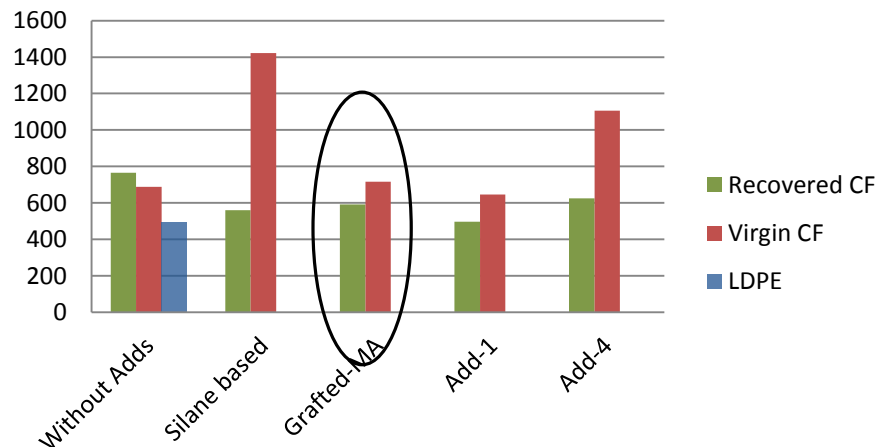
## Maximum Flexural Strength (MPa)



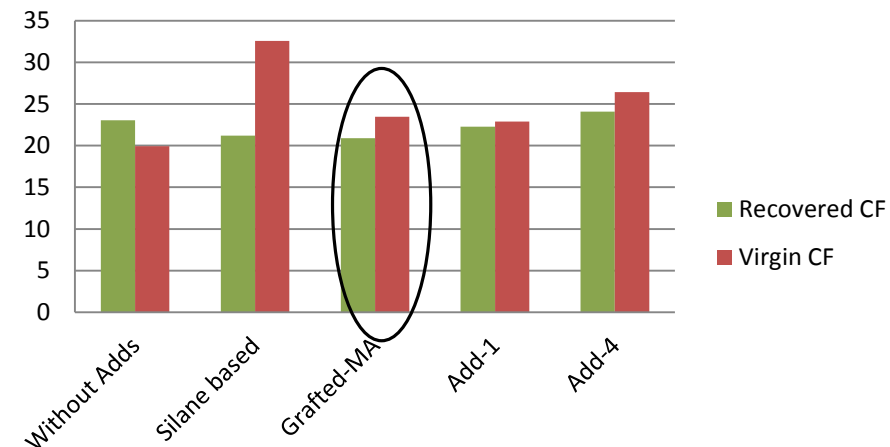
## Load at Max. Flexural Extension(N)



## Flexural Modulus (MPa)



## Impact Energy (J/m2)



- ❖ Using water alone, a maximum of 49 wt% of resin was removed
- ❖ With EG alone, 92% of resin removal was reached, higher percentages were reached by using EG/water mixture
- ❖ The liquid residuals from EG-promoted process could be valorised into fuel gas by supercritical water gasification or refined into liquid fuel
- ❖ LDPE's mechanical properties were improved when reinforced with RCF
- ❖ The commercial additive (silane-based) did not work well with the RCF; maybe due to loss of surface active groups compared to VCF

# Acknowledgements

Royal Society for International Exchanges Grant with University of Pannonia, Hungary

Dr Eyup Yildirir  
Dr Norbert Miskolczi  
Professor Paul T. Williams





Thank you for your  
attention

