

# SD30 Safety Data Sheet

## Hydrated Lime Range



Prepared in accordance with Annex II of the REACH Regulation EC 1907/2006,  
Regulation (EC) 1272/2008 and Regulation (EC) 453/2010

Revision date: March 2013

Printing Date: October 19, 2018

### SECTION 1. IDENTIFICATION OF THE SUBSTANCE/MIXTURE AND OF THE COMPANY

#### 1.1 Product identifier

Substance name: Hydrated lime, Calcium dihydroxide  
Synonyms: Slaked lime, Air slaked lime, Building lime, Fat lime, Chemical lime, Finishing lime, Mason's lime, Calcium dihydroxide, Calcium hydroxide, Calcium hydrate, Lime, Lime water  
Chemical name and formula: Calcium dihydroxide –  $\text{Ca}(\text{OH})_2$   
Trade name: Ultralime® Hydrated Lime, Biolime® H90  
CAS: 1305-62-0  
EINECS: 215-137-3  
Molecular Weight: 74.09 g/mol  
REACH Registration number: 01-2119475151-45-0019

#### 1.2 Relevant identified uses of the substance or mixture and uses advised against

**Use of the substance:** The substance is intended for the following non-exhaustive list of uses: Building material industry, Chemical industry, Agriculture, Biocidal use, Environmental protection (e.g. flue gas treatment, waste water treatment, sludge treatment), Drinking water treatment, Feed, food and pharmaceutical industry, Civil engineering, Paper and paint industry

##### 1.2.1 Identified uses

All uses listed in table 1 of the Appendix of this SDS are identified uses.

##### 1.2.2 Uses advised against

No use identified in Table 1 of the Appendix of this SDS is advised against.

#### 1.3 Details of the supplier of the safety data sheet

Name: Singleton Birch Limited  
Address: Melton Ross Quarries, Barnetby,  
North Lincolnshire DN38 6AE  
Phone N°: +44(0)1652 686000  
Fax N°: +44(0)1652 686081  
E-mail of competent person responsible for SDS in the MS or in the EU: kb@singletonbirch.co.uk; jt@singletonbirch.co.uk

#### 1.4 Emergency telephone number

European Emergency N°: 112  
National centre for Prevention & Treatment of Intoxications N°: National Chemicals Emergency Centre  
(NCEC) +44 (0) 870 190 6621

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Registered Number 9433 England  
Established 1815



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Emergency telephone at the company: +44(0)1652 686000 (24 hours)  
Available outside office hours: Yes

### SECTION 2. HAZARDS IDENTIFICATION

#### 2.1 Classification of the substance

##### 2.1.1 Classification according to Regulation (EC) 1272/2008

STOT Single Exp. 3, H335 Route of exposure: Inhalation

Skin Irritation 2, H315

Eye Damage 1, H318

##### 2.1.2 Classification according to Directive 67/548/EEC

Xi – irritant: R37, R38, R41

##### 2.1.3 Additional information

For full text of H-statements and R-phrases: see SECTION 16

#### 2.2 Label elements

##### 2.2.1 Labelling according to Regulation (EC) 1272/2008

Signal word:

Danger

Hazard pictogram:



#### Hazard statements:

H315: Causes skin irritation  
H318: Causes serious eye damage  
H335: May cause respiratory irritation

#### Precautionary statements:

P102: Keep out of reach of children  
P280: Wear protective gloves/protective clothing/eye protection/face protection  
P305+P351+P310: IF IN EYES: Rinse cautiously with water for several minutes. Immediately call a POISON CENTRE or doctor/physician  
P302+P352: IF ON SKIN: Wash with plenty of water  
P261: Avoid breathing dust/spray  
P304+P340: IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing  
P501: Dispose of contents/container in accordance with local, regional, national and international regulation – use a registered

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hazardous waste carrier/licence holder,  
and/or contact the manufacturer

### 2.3 Other hazards

The substance does not meet the criteria for PBT or vPvB substance.

No other hazards identified.

## SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS

### 3.1 Substances

#### Main constituent

CAS number	EC number	Registration No	Identification name	Weight % content (or range)	Classification according to 67/548/EEC
1305-62-0	215-137-3	01-2119475151-45-0019	Calcium hydroxide	>90%	Xi: R37,R38, R41

CAS number	EC number	Registration No	Identification name	Weight % content (or range)	Classification according to Regulation (EC) No 1272/2008 [CLP]
1305-62-0	215-137-3	01-2119475151-45-0019	Calcium hydroxide	>90%	Eye Dam 1 H318 Skin Irrit. 2 H315 STOT SE 3 (inhalation) H335

## SECTION 4. FIRST AID MEASURES

### 4.1 Description of first aid measures

#### General advice

No known delayed effects. Consult a physician for all exposures except for minor instances.

#### Following inhalation

Move source of dust or move person to fresh air. Obtain medical attention immediately.

#### Following skin contact

Carefully and gently brush the contaminated body surfaces in order to remove all traces of product. Wash affected area immediately with plenty of water. Remove contaminated clothing. If necessary seek medical advice.

#### Following eye contact

Rinse eyes immediately with plenty of water and seek medical advice.

#### After ingestion

Clean mouth with water and drink afterwards plenty of water. Do **NOT** induce vomiting. Obtain medical attention.

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### 4.2 Most important symptoms and effects, both acute and delayed

Calcium dihydroxide is not acutely toxic via the oral, dermal, or inhalation route. The substance is classified as irritating to skin and the respiratory tract, and entails a risk of serious damage to the eye. There is no concern for adverse systemic effects because local effects (pH-effect) are the major health hazard.

### 4.3 Indication of any immediate medical attention and special treatment needed

Follow the advises given in section 4.1

## SECTION 5. FIREFIGHTING MEASURES

### 5.1 Extinguishing media

#### 5.1.1 Suitable extinguishing media

Suitable extinguishing media: The product is not combustible. Use a dry powder, foam or CO<sub>2</sub> fire extinguisher to extinguish the surrounding fire. Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

#### 5.1.2 Unsuitable extinguishing media

Do not use water

### 5.2 Special hazards arising from the substance or mixture

None

### 5.3 Advice for fire fighters

Avoid generation of dust. Use breathing apparatus. Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

## SECTION 6. ACCIDENTAL RELEASE MEASURES

### 6.1 Personal precautions, protective equipment and emergency procedures

#### 6.1.1 For non-emergency personnel

Ensure adequate ventilation.

Keep dust levels to a minimum.

Keep unprotected persons away.

Avoid contact with skin, eyes, and clothing – wear suitable protective equipment (see section 8).

Avoid inhalation of dust – ensure that sufficient ventilation or suitable respiratory protective equipment is used, wear suitable protective equipment (see section 8).

#### 6.1.2 For emergency responders

Keep dust levels to a minimum.

Ensure adequate ventilation.

Keep unprotected persons away.

Avoid contact with skin, eyes, and clothing – wear suitable protective equipment (see section 8).

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Avoid inhalation of dust – ensure that sufficient ventilation or suitable respiratory protective equipment is used, wear suitable protective equipment (see section 8).

### 6.2 Environmental precautions

Contain the spillage. Keep the material dry if possible. Cover area if possible to avoid unnecessary dust hazard. Avoid uncontrolled spills to watercourses and drains (pH increase). Any large spillage into watercourses must be alerted to the Environment Agency or other regulatory body.

### 6.3 Methods and material for containment and cleaning up

In all cases avoid dust formation.

Keep the material dry if possible.

Pick up the product mechanically in a dry way.

Use vacuum suction unit, or shovel into bags.

### 6.4 Reference to other sections

For more information on exposure controls/personal protection or disposal considerations, please check section 8 and 13 and the Annex of this safety data sheet.

## SECTION 7. HANDLING AND STORAGE

### 7.1 Precautions for safe handling

#### 7.1.1 Protective measures

Avoid contact with skin and eyes. Wear protective equipment (refer to section 8 of this safety data sheet). Do not wear contact lenses when handling this product. It is also advisable to have individual pocket eyewash. Keep dust levels to a minimum. Minimize dust generation. Enclose dust sources, use exhaust ventilation (dust collector at handling points). Handling systems should preferably be enclosed. When handling bags usual precautions should be paid to the risks outlined in the Council Directive 90/269/EEC.

#### 7.1.2 Advice on general occupational hygiene

Avoid inhalation or ingestion and contact with skin and eyes. General occupational hygiene measures are required to ensure safe handling of the substance. These measures involve good personal and housekeeping practices (i.e. regular cleaning with suitable cleaning devices), no drinking, eating and smoking at the workplace. Shower and change clothes at end of work shift. Do not wear contaminated clothing at home.

### 7.2 Conditions for safe storage, including any incompatibilities

The substance should be stored under dry conditions. Any contact with air and moisture should be avoided. Bulk storage should be in purpose – designed silos. Keep away from acids, significant quantities of paper, straw, and nitro compounds. Keep out of reach of children. Do not use aluminium for transport or storage if there is a risk of contact with water.

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### 7.3 Specific end use(s)

Please check the identified uses in table 1 of the Appendix of this SDS.

For more information please see the relevant exposure scenario, available via your supplier/given in the Appendix, and check section 2.1: Control of worker exposure.

## SECTION 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

### 8.1 Control parameters

#### DNEL's

Workers				
Route of exposure	Acute effect local	Acute effects systemic	Chronic effects local	Chronic effects systemic
Oral	Not required			
Inhalation	4 mg / m <sup>3</sup> (Respirable dust)	No hazard identified	1 mg / m <sup>3</sup> (Respirable dust)	No hazard identified
Dermal	Hazard identified but no DNEL available	No hazard identified	Hazard identified but no DNEL available	No hazard identified

Consumers				
Route of exposure	Acute effect local	Acute effects systemic	Chronic effects local	Chronic effects systemic
Oral	No exposure expected	No hazard identified	No exposure expected	No hazard identified
Inhalation	4 mg / m <sup>3</sup> (Respirable dust)	No hazard identified	1 mg / m <sup>3</sup> (Respirable dust)	No hazard identified
Dermal	Hazard identified but no DNEL available	No hazard identified	Hazard identified but no DNEL available	No hazard identified

#### PNEC's

Environment protection target	PNEC	Remarks
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Fresh water	0.49 mg / L	
Freshwater sediments	No PNEC available	Insufficient data available
Marine water	0.32 mg / L	
Marine sediments	No PNEC available	Insufficient data available
Food (bioaccumulation)	No hazard identified	No potential for bioaccumulation
Microorganisms in sewage treatment	3 mg / L	
Soil (agricultural)	1080 mg / kg soil dw	
Air	No hazard identified	

**Workplace Exposure Limit (WEL), 8 h TWA:** 5 mg/m<sup>3</sup>

**Occupational Exposure Limit (OEL), 8h TWA:** 1 mg/m<sup>3</sup> respirable dust of calcium oxide

**Short-term exposure limit (STEL), 15 min:** 4 mg/m<sup>3</sup> respirable dust of calcium oxide

### 8.2 Exposure controls

To control potential exposures, generation of dust should be avoided. Further, appropriate protective equipment is recommended. Eye protection equipment (e.g. goggles or visors) must be worn, unless potential contact with the eye can be excluded by the nature and type of application (i.e. closed process). Additionally, face protection, protective clothing and safety shoes are required to be worn as appropriate.

Please check the relevant exposure scenario, given in the Appendix/available via your supplier.

#### 8.2.1 Appropriate engineering controls

If user operations generate dust, use process enclosures, local exhaust ventilation, or other engineering controls to keep airborne dust levels below recommended exposure limits.

#### 8.2.2 Individual protection measures, such as personal protective equipment

##### 8.2.2.1 Eye/face protection

Do not wear contact lenses. For powders, tight fitting goggles with side shields, or wide vision full goggles. Eyewash facilities should be readily available.

##### 8.2.2.2 Skin protection

Since calcium dihydroxide is classified as irritating to skin, dermal exposure has to be minimised as far as technically feasible. The use of protective gloves (nitrile), protective standard working clothes

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fully covering skin, full length trousers, long sleeved overalls, with close fittings at openings and shoes resistant to caustics and avoiding dust penetration are required to be worn.

### 8.2.2.3 Respiratory protection

Local ventilation to keep levels below established threshold values is recommended. A suitable particle filter mask is recommended, depending on the expected exposure levels - please check the relevant exposure scenario, given in the Appendix/available via your supplier.

### 8.2.2.4 Thermal hazards

The substance does not represent a thermal hazard, thus special consideration is not required.

### 8.2.3 Environmental exposure controls

All ventilation systems should be filtered before discharge to atmosphere.

Avoid releasing to the environment.

Contain the spillage. Any large spillage into watercourses must be alerted to the regulatory authority responsible for environmental protection or other regulatory body.

For detailed explanations of the risk management measures that adequately control exposure of the environment to the substance please check the relevant exposure scenario, available via your supplier.

For further detailed information, please check the Appendix of this SDS.

## SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

### 9.1 Information on basic physical and chemical properties

Appearance:	White or off white (beige) fine powder
Odour:	odourless
Odour threshold:	not applicable
pH:	12.4 (saturated solution at 20 °C)
Melting point:	> 450 °C (study result, EU A.1 method)
Boiling point:	not applicable (solid with a melting point > 450 °C)
Flash point:	not applicable (solid with a melting point > 450 °C)
Evaporation rate:	not applicable (solid with a melting point > 450 °C)
Flammability:	non flammable (study result, EU A.10 method)
Explosive limits:	non explosive (void of any chemical structures commonly associated with explosive properties)
Vapour pressure:	not applicable (solid with a melting point > 450 °C)
Vapour density:	not applicable
Relative density:	2.24 (study result, EU A.3 method)
Solubility in water:	1844.9 mg/L (study results, EU A.6 method)
Partition coefficient:	not applicable (inorganic substance)

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Auto ignition temperature:	no relative self-ignition temperature below 400 °C (study result, EU A.16 method)
Decomposition temperature:	When heated above 580°C, calcium dihydroxide decomposes to produce calcium oxide (CaO) and water (H <sub>2</sub> O)
Viscosity:	not applicable (solid with a melting point > 450 °C)
Oxidising properties:	no oxidising properties (Based on the chemical structure, the substance does not contain a surplus of oxygen or any structural groups known to be correlated with a tendency to react exothermally with combustible material)

### 9.2 Other information

Not available

## SECTION 10. STABILITY AND REACTIVITY

### 10.1 Reactivity

In aqueous media Ca(OH)<sub>2</sub> dissociates resulting in the formation of calcium cations and hydroxyl anions (when below the limit of water solubility).

### 10.2 Chemical stability

Under normal conditions of use and storage, calcium dihydroxide is stable.

### 10.3 Possibility of hazardous reactions

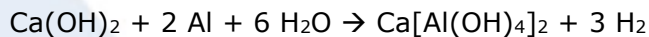
Calcium dihydroxide reacts exothermically with acids. When heated above 580 °C, calcium dihydroxide decomposes to produce calcium oxide (CaO) and water (H<sub>2</sub>O):  $\text{Ca(OH)}_2 \rightarrow \text{CaO} + \text{H}_2\text{O}$ . Calcium oxide reacts with water and generates heat. This may cause risk to flammable material.

### 10.4 Conditions to avoid

Minimise exposure to air and moisture to avoid degradation.

### 10.5 Incompatible materials

Calcium dihydroxide reacts exothermically with acids to form salts. Calcium dihydroxide reacts with aluminium and brass in the presence of moisture leading to the production of hydrogen.



### 10.6 Hazardous decomposition products

None

Further information: calcium hydroxide reacts with carbon dioxide to form calcium carbonate, which is a common material in nature.

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### SECTION 11. TOXICOLOGICAL INFORMATION

#### 11.1 Information on toxicological effects

##### a. Acute toxicity

Oral LD50 > 2000 mg/kg bw (OECD 425, rat)  
Dermal LD50 > 2500 mg/kg bw (OECD 402, rabbit)  
Inhalation no data available  
Calcium hydroxide is not acutely toxic.  
Classification for acute toxicity is not warranted.

##### b. Skin corrosion/irritation

Calcium hydroxide is irritating to skin (in vivo, rabbit).  
Based on experimental results, calcium hydroxide requires classification as irritating to skin [R38, irritating to skin; Skin Irrit 2 (H315 - Causes skin irritation)].

##### c. Serious eye damage/irritation

Calcium hydroxide entails a risk of serious damage to the eye (eye irritation studies (in vivo, rabbit).  
Based on experimental results, calcium hydroxide requires classification as severely irritating to the eye [R41, Risk of serious damage to eye; Eye Damage 1 (H318 - Causes serious eye damage)].

##### d. Respiratory or skin sensitisation

No data available.  
Calcium hydroxide is considered not to be a skin sensitiser, based on the nature of the effect (pH shift) and the essential requirement of calcium for human nutrition.  
Classification for sensitisation is not warranted.

##### e. Germ cell mutagenicity

Bacterial reverse mutation assay (Ames test, OECD 471): Negative  
Mammalian chromosome aberration test: Negative  
In view of the omnipresence and essentiality of Ca and of the physiological non-relevance of any pH shift induced by lime in aqueous media, lime is obviously void of any genotoxic potential, including germ cell mutagenicity.  
Classification for genotoxicity is not warranted.

##### f. Carcinogenicity

Calcium (administered as Ca-lactate) is not carcinogenic (experimental result, rat).  
The pH effect of calcium hydroxide does not give rise to a carcinogenic risk.  
Human epidemiological data support lack of any carcinogenic potential of calcium hydroxide.  
Classification for carcinogenicity is not warranted.

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### g. Reproductive toxicity

Calcium (administered as Ca-carbonate) is not toxic to reproduction (experimental result, mouse). The pH effect does not give rise to a reproductive risk.

Human epidemiological data support lack of any potential for reproductive toxicity of calcium hydroxide.

Both in animal studies and human clinical studies on various calcium salts no reproductive or developmental effects were detected. Also see the Scientific Committee on Food (Section 16.6). Thus, calcium hydroxide is not toxic for reproduction and/or development.

Classification for reproductive toxicity according to regulation (EC) 1272/2008 is not required.

### h. STOT-single exposure

From human data it is concluded that  $\text{Ca}(\text{OH})_2$  is irritating to the respiratory tract. As summarised and evaluated in the SCOEL recommendation (Anonymous, 2008), based on human data calcium hydroxide is classified as irritating to the respiratory system [R37, Irritating to respiratory system; STOT SE 3 (H335 – May cause respiratory irritation)].

### i. STOT-repeated exposure

Toxicity of calcium via the oral route is addressed by upper intake levels (UL) for adults determined by the Scientific Committee on Food (SCF), being

UL = 2500 mg/d, corresponding to 36 mg/kg bw/d (70 kg person) for calcium.

Toxicity of  $\text{Ca}(\text{OH})_2$  via the dermal route is not considered as relevant in view of the anticipated insignificant absorption through skin and due to local irritation as the primary health effect (pH shift).

Toxicity of  $\text{Ca}(\text{OH})_2$  via inhalation (local effect, irritation of mucous membranes) is addressed by an 8-h TWA determined by the Scientific Committee on Occupational Exposure Limits (SCOEL) of 1 mg/m<sup>3</sup> respirable dust (see Section 8.1).

Therefore, classification of  $\text{Ca}(\text{OH})_2$  for toxicity upon prolonged exposure is not required.

### j. Aspiration hazard

Calcium hydroxide is not known to present an aspiration hazard.

## SECTION 12. ECOLOGICAL INFORMATION

### 12.1 Toxicity

#### 12.1.1 Acute/Prolonged toxicity to fish

LC<sub>50</sub> (96h) for freshwater fish: 50.6 mg/l

LC<sub>50</sub> (96h) for marine water fish: 457 mg/l

#### 12.1.2 Acute/Prolonged toxicity to aquatic invertebrates

EC<sub>50</sub> (48h) for freshwater invertebrates: 49.1 mg/l

LC<sub>50</sub> (96h) for marine water invertebrates: 158 mg/l

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# SD30 Safety Data Sheet

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### 12.1.3 Acute/Prolonged toxicity to aquatic plants

EC<sub>50</sub> (72h) for freshwater algae: 184.57 mg/l

NOEC (72h) for freshwater algae: 48 mg/l

### 12.1.4 Toxicity to micro-organisms e.g. bacteria

At high concentration, through the rise of temperature and pH, calcium dihydroxide is used for disinfection of sewage sludges

### 12.1.5 Chronic toxicity to aquatic organisms

NOEC (14d) for marine water invertebrates: 32 mg/l

### 12.1.6 Toxicity to soil dwelling organisms

EC<sub>10</sub>/LC<sub>10</sub> or NOEC for soil macro organisms: 2000 mg/kg soil dw

EC<sub>10</sub>/LC<sub>10</sub> or NOEC for soil micro organisms: 12000 mg/kg soil dw

### 12.1.7 Toxicity to terrestrial plants

NOEC (21d) for terrestrial plants: 1080 mg/kg

### 12.1.8 General effect

Acute pH-effect. Although this product is useful to correct water acidity, an excess of more than 1 g/l may be harmful to aquatic life. pH-value of > 12 will rapidly decrease as result of dilution and carbonation.

### 12.2 Persistence and degradability

Not relevant for inorganic substances

### 12.3 Bioaccumulative potential

Not relevant for inorganic substances

### 12.4 Mobility in soil

Calcium dihydroxide which is sparingly soluble, and present a low mobility in most soils

### 12.5 Results of PBT and vPvB assessment

Not relevant for inorganic substances

### 12.6 Other adverse effects

No other adverse effects are identified

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### SECTION 13. DISPOSAL CONSIDERATIONS

#### 13.1 Waste treatment methods

Disposal of calcium dihydroxide should be in accordance with local and national legislation.

Processing, use or contamination of this product may change the waste management options.

Dispose of container and unused contents in accordance with applicable member state and local requirements.

The used packing is only meant for packing this product; it should not be reused for other purposes. After usage, empty the packing completely.

### SECTION 14. TRANSPORT INFORMATION

Calcium hydroxide is not classified as hazardous for transport (ADR (Road), RID (Rail), ICAO/IATA (air), AND (inland waterways) and IMDG (Sea)).

#### 14.1 UN-Number

Not regulated

#### 14.2 UN proper shipping name

Not regulated

#### 14.3 Transport hazard class

Not regulated

#### 14.4 Packing group

Not regulated

#### 14.5 Environmental hazards

None

#### 14.6 Special precautions for user

Avoid any release of dust during transportation, by using air-tight tanks

#### 14.7 Transport in bulk according to Annex II of MARPOL73/78 and the IBC Code

Not regulated.

### SECTION 15. REGULATORY INFORMATION

#### 15.1 Safety, health and environmental regulations/legislation specific for the substance

Authorisations: Not required

Restrictions on use: None

Other EU regulations: Calcium hydroxide is not a SEVESO substance, not an ozone depleting substance and not a persistent organic pollutant.

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National regulations: Water endangering class 1 (Germany)

### 15.2 Chemical safety assessment

A chemical safety assessment has been carried out for this substance.

### SECTION 16. OTHER INFORMATION

Data are based on our latest knowledge but do not constitute a guarantee for any specific product features and do not establish a legally valid contractual relationship.

#### 16.1 Hazard Statements

H315: Causes skin irritation  
H318: Causes serious eye damage  
H335: May cause respiratory irritation

#### 16.2 Precautionary Statements

P102: Keep out of reach of children  
P280: Wear protective gloves/protective clothing/eye protection/face protection  
P305+P351: IF IN EYES: Rinse cautiously with water for several minutes  
P310: Immediately call a POISON CENTRE or doctor/physician  
P302+P352: IF ON SKIN: Wash with plenty of soap and water  
P261: Avoid breathing dust/fume/gas/mist/vapours/spray  
P304+P340: IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing  
P501: Dispose of contents/container in accordance with local/regional/national/international regulation - use a registered hazardous waste carrier/licence holder, and/or contact the manufacturer

#### 16.3 Risk Phrases

R37: Irritating to respiratory system  
R38: Irritating to skin  
R41: Risk of serious damage to eyes

#### 16.4 Safety Phrases

S2: Keep out of the reach of children  
S25: Avoid contact with eyes  
S26: In case of contact with eyes, rinse immediately with plenty of water and seek medical advice  
S37: Wear suitable gloves  
S39: Wear eye/face protection

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### 16.5 Abbreviations

EC<sub>50</sub>: median effective concentration  
LC<sub>50</sub>: median lethal concentration  
LD<sub>50</sub>: median lethal dose  
NOEC: no observable effect concentration  
WEL: workplace exposure limit  
OEL: occupational exposure limit  
PBT: persistent, bioaccumulative, toxic chemical  
PNEC: predicted no-effect concentration  
STEL: short-term exposure limit  
TWA: time weighted average  
vPvB: very persistent, very bioaccumulative chemical  
EULA: European Lime Association  
DNEL: derived no-effect level

### 16.6 Key literature references

Anonymous, 2006: Tolerable upper intake levels for vitamins and minerals Scientific Committee on Food, European Food Safety Authority, ISBN: 92-9199-014-0 [SCF document] Anonymous, 2008: Recommendation from the Scientific Committee on Occupational Exposure Limits (SCOEL) for calcium oxide (CaO) and calcium hydroxide (Ca(OH)<sub>2</sub>), European Commission, DG Employment, Social Affairs and Equal Opportunities, SCOEL/SUM/137 February 2008.

### 16.7 Revision

SDS revised in accordance with EULA SDS format  
The word "SECTION" has been included in each chapter.  
Section 1.2 Relevant identified uses of the substance and uses advised against amended.  
Section 2.1.1 Classification of the substance – 'H' numbers added.  
Section 2.1.2 Classification of the substance – 'R' numbers added.  
Section 2.1.3 'Additional information' added.  
Section 2.2.2 'Labelling according to Directive 67/548/EEC' removed.  
Section 3.2 'Hazardous ingredients' tabulised.  
Section 4, 4.1 Description of first aid measures, Following ingestion – word 'following' changed to 'after' ingestion  
Section 8, 8.1 Control parameters, DNEL's, Oral 'No exposure expected' changed to 'Not required'. Information tabulised.  
Section 8, 8.2.2.1 Eye/Face protection - It is also advisable to have individual pocket eyewash.' replaced with 'Eyewash facilities should be readily available.'  
Section 11, 11.1 Information on toxicological effects amended.  
Section 14: TRANSPORT INFORMATION - 'The mixture is not classified as hazardous for transport (ADR (Road), RID (Rail), IMDG / GGVSea (Sea)).' replaced with 'Calcium hydroxide is not classified as hazardous for transport [ADR (Road), RID (Rail), ICAO/IATA (air), AND (inland waterways) and IMDG (Sea)].'.  
Section 16, 16.6 Key literature references – removed.

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### Disclaimer

This safety data sheet (SDS) is based on the legal provisions of the REACH Regulation (EC 1907/2006; article 31 and Annex II), as amended. Its contents are intended as a guide to the appropriate precautionary handling of the material. It is the responsibility of recipients of this SDS to ensure that the information contained therein is properly read and understood by all people who may use, handle, dispose or in any way come in contact with the product. Information and instructions provided in this SDS are based on the current state of scientific and technical knowledge at the date of issue indicated. It should not be construed as any guarantee of technical performance, suitability for particular applications, and does not establish a legally valid contractual relationship. This version of the SDS supersedes all previous versions.

### **ANNEX**

Addition of exposure Scenarios as applicable - Please see Appendix SD30A SDS – Hydrate Lime Range.

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